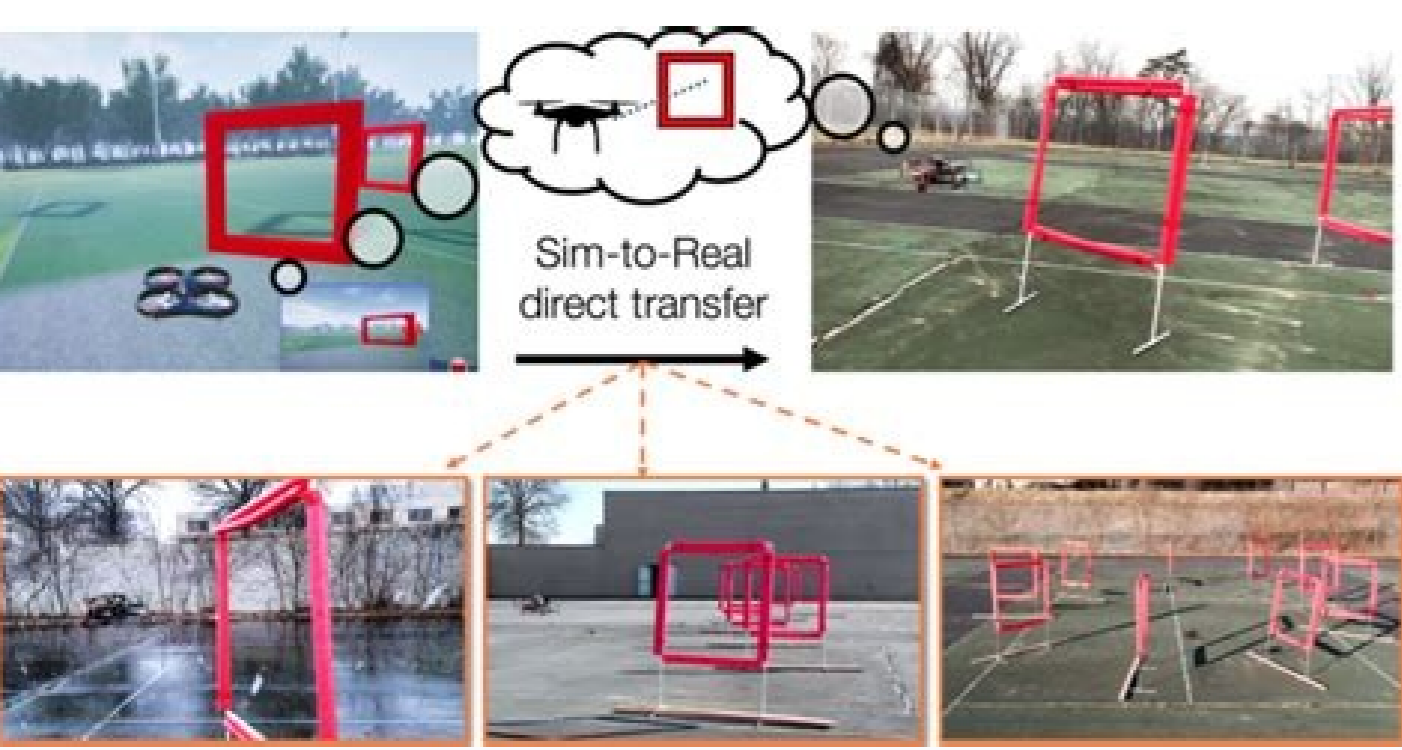


I'm not robot!



Problem #5: How many grams of pirazol (C3H4N2) would have to be dissolved in 736 g of sword to decrease the freezing point by 15.0 ° C? The constant freezing point for sword is 40. ° C/m. Solution: $\Delta T = i \cdot k_f \cdot m$ $15.0 = i \cdot 1.86 \cdot m$ $m = 15.0 / (1.86 \cdot i)$ $m = 8.06$ $m = 8.06 \cdot 736 \text{ g sword} = 5952.16 \text{ g sword}$

Problem 6: which is the freezing point of a solution Prepared by adding 140. G Triquecin (C19H24O5) to 0.746 kg of benzene? The freezing point of pure benzene is 5.5 ° C. The constant freezing point for benzene is 5.12 ° C/m. Solution: $\Delta T = i \cdot k_f \cdot m$ $5.12 = i \cdot 5.12 \cdot m$ $m = 5.12 / (5.12 \cdot i)$ $m = 1$ $m = 1 \cdot 140 \text{ g} / 0.746 \text{ kg} = 187.67 \text{ g/kg}$

Problem #7: What is the freezing point depression when 309 g of isoprene (C8H16) is dissolved in 747 g of ethil is having? The constant freezing point for the ethical is 1.79 ° C/m. Solution: $309 \text{ g} / 68.1182 \text{ g/mol} = 4.536233 \text{ mol}$ $\Delta T = i \cdot k_f \cdot m$ $\Delta T = 1 \cdot 1.79 \cdot 4.536233 = 8.11$ $T = 1.79 - 8.11 = -6.32 \text{ °C}$

Problem #8: What is the freezing point of a prepared solution by adding 239.0 g of copper sulfate penthydrate (II) to 4.00 liters of water? The depression of the freezing point of the water is 1.86 ° C/m. Solution: $239.0 \text{ g} / 249.681 \text{ g/mol} = 0.95722 \text{ mol}$ $\Delta T = i \cdot k_f \cdot m$ $\Delta T = 1 \cdot 1.86 \cdot 0.95722 = 1.78$ $T = 0 - 1.78 = -1.78 \text{ °C}$

Problem #9: A solution containing ascance (vitamin C) at 250. g of water freezes to 4 ° C. Calculate molar mass (in g/mol) of the solute. Solution: 1) Use $\Delta T = i \cdot k_f \cdot m$ $4 = 1 \cdot 1.86 \cdot m$ $m = 4 / 1.86 = 2.15$ $m = 2.15 \text{ mol} / 0.250 \text{ kg} = 8.6 \text{ g/mol}$

2) determine how many moles of the compound dissolve: $n = i \cdot k_f \cdot m / \Delta T = 1 \cdot 1.86 \cdot 2.15 / 4 = 0.99$ $n = 0.99 \text{ mol}$

3) determine the molar mass: $M = 239.0 \text{ g} / 0.0279 \text{ mol} = 8566.31 \text{ g/mol}$

4) determine the molar mass of the compound: $M = 8566.31 \text{ g/mol} - 239.0 \text{ g} = 8327.31 \text{ g/mol}$

5) determine the molar mass of the compound: $M = 8327.31 \text{ g/mol} - 239.0 \text{ g} = 8088.31 \text{ g/mol}$

6) determine the molar mass of the compound: $M = 8088.31 \text{ g/mol} - 239.0 \text{ g} = 7849.31 \text{ g/mol}$

7) determine the molar mass of the compound: $M = 7849.31 \text{ g/mol} - 239.0 \text{ g} = 7610.31 \text{ g/mol}$

8) determine the molar mass of the compound: $M = 7610.31 \text{ g/mol} - 239.0 \text{ g} = 7371.31 \text{ g/mol}$

9) determine the molar mass of the compound: $M = 7371.31 \text{ g/mol} - 239.0 \text{ g} = 7132.31 \text{ g/mol}$

10) determine the molar mass of the compound: $M = 7132.31 \text{ g/mol} - 239.0 \text{ g} = 6893.31 \text{ g/mol}$

11) determine the molar mass of the compound: $M = 6893.31 \text{ g/mol} - 239.0 \text{ g} = 6654.31 \text{ g/mol}$

12) determine the molar mass of the compound: $M = 6654.31 \text{ g/mol} - 239.0 \text{ g} = 6415.31 \text{ g/mol}$

13) determine the molar mass of the compound: $M = 6415.31 \text{ g/mol} - 239.0 \text{ g} = 6176.31 \text{ g/mol}$

14) determine the molar mass of the compound: $M = 6176.31 \text{ g/mol} - 239.0 \text{ g} = 5937.31 \text{ g/mol}$

15) determine the molar mass of the compound: $M = 5937.31 \text{ g/mol} - 239.0 \text{ g} = 5698.31 \text{ g/mol}$

16) determine the molar mass of the compound: $M = 5698.31 \text{ g/mol} - 239.0 \text{ g} = 5459.31 \text{ g/mol}$

17) determine the molar mass of the compound: $M = 5459.31 \text{ g/mol} - 239.0 \text{ g} = 5220.31 \text{ g/mol}$

18) determine the molar mass of the compound: $M = 5220.31 \text{ g/mol} - 239.0 \text{ g} = 4981.31 \text{ g/mol}$

19) determine the molar mass of the compound: $M = 4981.31 \text{ g/mol} - 239.0 \text{ g} = 4742.31 \text{ g/mol}$

20) determine the molar mass of the compound: $M = 4742.31 \text{ g/mol} - 239.0 \text{ g} = 4503.31 \text{ g/mol}$

21) determine the molar mass of the compound: $M = 4503.31 \text{ g/mol} - 239.0 \text{ g} = 4264.31 \text{ g/mol}$

22) determine the molar mass of the compound: $M = 4264.31 \text{ g/mol} - 239.0 \text{ g} = 4025.31 \text{ g/mol}$

23) determine the molar mass of the compound: $M = 4025.31 \text{ g/mol} - 239.0 \text{ g} = 3786.31 \text{ g/mol}$

24) determine the molar mass of the compound: $M = 3786.31 \text{ g/mol} - 239.0 \text{ g} = 3547.31 \text{ g/mol}$

25) determine the molar mass of the compound: $M = 3547.31 \text{ g/mol} - 239.0 \text{ g} = 3308.31 \text{ g/mol}$

26) determine the molar mass of the compound: $M = 3308.31 \text{ g/mol} - 239.0 \text{ g} = 3069.31 \text{ g/mol}$

27) determine the molar mass of the compound: $M = 3069.31 \text{ g/mol} - 239.0 \text{ g} = 2830.31 \text{ g/mol}$

28) determine the molar mass of the compound: $M = 2830.31 \text{ g/mol} - 239.0 \text{ g} = 2591.31 \text{ g/mol}$

29) determine the molar mass of the compound: $M = 2591.31 \text{ g/mol} - 239.0 \text{ g} = 2352.31 \text{ g/mol}$

30) determine the molar mass of the compound: $M = 2352.31 \text{ g/mol} - 239.0 \text{ g} = 2113.31 \text{ g/mol}$

31) determine the molar mass of the compound: $M = 2113.31 \text{ g/mol} - 239.0 \text{ g} = 1874.31 \text{ g/mol}$

32) determine the molar mass of the compound: $M = 1874.31 \text{ g/mol} - 239.0 \text{ g} = 1635.31 \text{ g/mol}$

33) determine the molar mass of the compound: $M = 1635.31 \text{ g/mol} - 239.0 \text{ g} = 1396.31 \text{ g/mol}$

34) determine the molar mass of the compound: $M = 1396.31 \text{ g/mol} - 239.0 \text{ g} = 1157.31 \text{ g/mol}$

35) determine the molar mass of the compound: $M = 1157.31 \text{ g/mol} - 239.0 \text{ g} = 918.31 \text{ g/mol}$

36) determine the molar mass of the compound: $M = 918.31 \text{ g/mol} - 239.0 \text{ g} = 679.31 \text{ g/mol}$

37) determine the molar mass of the compound: $M = 679.31 \text{ g/mol} - 239.0 \text{ g} = 440.31 \text{ g/mol}$

38) determine the molar mass of the compound: $M = 440.31 \text{ g/mol} - 239.0 \text{ g} = 201.31 \text{ g/mol}$

39) determine the molar mass of the compound: $M = 201.31 \text{ g/mol} - 239.0 \text{ g} = -38.69 \text{ g/mol}$

40) determine the molar mass of the compound: $M = -38.69 \text{ g/mol} - 239.0 \text{ g} = -277.69 \text{ g/mol}$

41) determine the molar mass of the compound: $M = -277.69 \text{ g/mol} - 239.0 \text{ g} = -516.69 \text{ g/mol}$

42) determine the molar mass of the compound: $M = -516.69 \text{ g/mol} - 239.0 \text{ g} = -755.69 \text{ g/mol}$

43) determine the molar mass of the compound: $M = -755.69 \text{ g/mol} - 239.0 \text{ g} = -994.69 \text{ g/mol}$

44) determine the molar mass of the compound: $M = -994.69 \text{ g/mol} - 239.0 \text{ g} = -1233.69 \text{ g/mol}$

45) determine the molar mass of the compound: $M = -1233.69 \text{ g/mol} - 239.0 \text{ g} = -1472.69 \text{ g/mol}$

46) determine the molar mass of the compound: $M = -1472.69 \text{ g/mol} - 239.0 \text{ g} = -1711.69 \text{ g/mol}$

47) determine the molar mass of the compound: $M = -1711.69 \text{ g/mol} - 239.0 \text{ g} = -1950.69 \text{ g/mol}$

48) determine the molar mass of the compound: $M = -1950.69 \text{ g/mol} - 239.0 \text{ g} = -2189.69 \text{ g/mol}$

49) determine the molar mass of the compound: $M = -2189.69 \text{ g/mol} - 239.0 \text{ g} = -2428.69 \text{ g/mol}$

50) determine the molar mass of the compound: $M = -2428.69 \text{ g/mol} - 239.0 \text{ g} = -2667.69 \text{ g/mol}$

51) determine the molar mass of the compound: $M = -2667.69 \text{ g/mol} - 239.0 \text{ g} = -2906.69 \text{ g/mol}$

52) determine the molar mass of the compound: $M = -2906.69 \text{ g/mol} - 239.0 \text{ g} = -3145.69 \text{ g/mol}$

53) determine the molar mass of the compound: $M = -3145.69 \text{ g/mol} - 239.0 \text{ g} = -3384.69 \text{ g/mol}$

54) determine the molar mass of the compound: $M = -3384.69 \text{ g/mol} - 239.0 \text{ g} = -3623.69 \text{ g/mol}$

55) determine the molar mass of the compound: $M = -3623.69 \text{ g/mol} - 239.0 \text{ g} = -3862.69 \text{ g/mol}$

56) determine the molar mass of the compound: $M = -3862.69 \text{ g/mol} - 239.0 \text{ g} = -4101.69 \text{ g/mol}$

57) determine the molar mass of the compound: $M = -4101.69 \text{ g/mol} - 239.0 \text{ g} = -4340.69 \text{ g/mol}$

58) determine the molar mass of the compound: $M = -4340.69 \text{ g/mol} - 239.0 \text{ g} = -4579.69 \text{ g/mol}$

59) determine the molar mass of the compound: $M = -4579.69 \text{ g/mol} - 239.0 \text{ g} = -4818.69 \text{ g/mol}$

60) determine the molar mass of the compound: $M = -4818.69 \text{ g/mol} - 239.0 \text{ g} = -5057.69 \text{ g/mol}$

61) determine the molar mass of the compound: $M = -5057.69 \text{ g/mol} - 239.0 \text{ g} = -5296.69 \text{ g/mol}$

62) determine the molar mass of the compound: $M = -5296.69 \text{ g/mol} - 239.0 \text{ g} = -5535.69 \text{ g/mol}$

63) determine the molar mass of the compound: $M = -5535.69 \text{ g/mol} - 239.0 \text{ g} = -5774.69 \text{ g/mol}$

64) determine the molar mass of the compound: $M = -5774.69 \text{ g/mol} - 239.0 \text{ g} = -6013.69 \text{ g/mol}$

65) determine the molar mass of the compound: $M = -6013.69 \text{ g/mol} - 239.0 \text{ g} = -6252.69 \text{ g/mol}$

66) determine the molar mass of the compound: $M = -6252.69 \text{ g/mol} - 239.0 \text{ g} = -6491.69 \text{ g/mol}$

67) determine the molar mass of the compound: $M = -6491.69 \text{ g/mol} - 239.0 \text{ g} = -6730.69 \text{ g/mol}$

68) determine the molar mass of the compound: $M = -6730.69 \text{ g/mol} - 239.0 \text{ g} = -6969.69 \text{ g/mol}$

69) determine the molar mass of the compound: $M = -6969.69 \text{ g/mol} - 239.0 \text{ g} = -7208.69 \text{ g/mol}$

70) determine the molar mass of the compound: $M = -7208.69 \text{ g/mol} - 239.0 \text{ g} = -7447.69 \text{ g/mol}$

71) determine the molar mass of the compound: $M = -7447.69 \text{ g/mol} - 239.0 \text{ g} = -7686.69 \text{ g/mol}$

72) determine the molar mass of the compound: $M = -7686.69 \text{ g/mol} - 239.0 \text{ g} = -7925.69 \text{ g/mol}$

73) determine the molar mass of the compound: $M = -7925.69 \text{ g/mol} - 239.0 \text{ g} = -8164.69 \text{ g/mol}$

74) determine the molar mass of the compound: $M = -8164.69 \text{ g/mol} - 239.0 \text{ g} = -8403.69 \text{ g/mol}$

75) determine the molar mass of the compound: $M = -8403.69 \text{ g/mol} - 239.0 \text{ g} = -8642.69 \text{ g/mol}$

76) determine the molar mass of the compound: $M = -8642.69 \text{ g/mol} - 239.0 \text{ g} = -8881.69 \text{ g/mol}$

77) determine the molar mass of the compound: $M = -8881.69 \text{ g/mol} - 239.0 \text{ g} = -9120.69 \text{ g/mol}$

78) determine the molar mass of the compound: $M = -9120.69 \text{ g/mol} - 239.0 \text{ g} = -9359.69 \text{ g/mol}$

79) determine the molar mass of the compound: $M = -9359.69 \text{ g/mol} - 239.0 \text{ g} = -9598.69 \text{ g/mol}$

80) determine the molar mass of the compound: $M = -9598.69 \text{ g/mol} - 239.0 \text{ g} = -9837.69 \text{ g/mol}$

81) determine the molar mass of the compound: $M = -9837.69 \text{ g/mol} - 239.0 \text{ g} = -10076.69 \text{ g/mol}$

82) determine the molar mass of the compound: $M = -10076.69 \text{ g/mol} - 239.0 \text{ g} = -10315.69 \text{ g/mol}$

83) determine the molar mass of the compound: $M = -10315.69 \text{ g/mol} - 239.0 \text{ g} = -10554.69 \text{ g/mol}$

84) determine the molar mass of the compound: $M = -10554.69 \text{ g/mol} - 239.0 \text{ g} = -10793.69 \text{ g/mol}$

85) determine the molar mass of the compound: $M = -10793.69 \text{ g/mol} - 239.0 \text{ g} = -11032.69 \text{ g/mol}$

86) determine the molar mass of the compound: $M = -11032.69 \text{ g/mol} - 239.0 \text{ g} = -11271.69 \text{ g/mol}$

87) determine the molar mass of the compound: $M = -11271.69 \text{ g/mol} - 239.0 \text{ g} = -11510.69 \text{ g/mol}$

88) determine the molar mass of the compound: $M = -11510.69 \text{ g/mol} - 239.0 \text{ g} = -11749.69 \text{ g/mol}$

89) determine the molar mass of the compound: $M = -11749.69 \text{ g/mol} - 239.0 \text{ g} = -11988.69 \text{ g/mol}$

90) determine the molar mass of the compound: $M = -11988.69 \text{ g/mol} - 239.0 \text{ g} = -12227.69 \text{ g/mol}$

91) determine the molar mass of the compound: $M = -12227.69 \text{ g/mol} - 239.0 \text{ g} = -12466.69 \text{ g/mol}$

92) determine the molar mass of the compound: $M = -12466.69 \text{ g/mol} - 239.0 \text{ g} = -12705.69 \text{ g/mol}$

93) determine the molar mass of the compound: $M = -12705.69 \text{ g/mol} - 239.0 \text{ g} = -12944.69 \text{ g/mol}$

94) determine the molar mass of the compound: $M = -12944.69 \text{ g/mol} - 239.0 \text{ g} = -13183.69 \text{ g/mol}$

95) determine the molar mass of the compound: $M = -13183.69 \text{ g/mol} - 239.0 \text{ g} = -13422.69 \text{ g/mol}$

96) determine the molar mass of the compound: $M = -13422.69 \text{ g/mol} - 239.0 \text{ g} = -13661.69 \text{ g/mol}$

97) determine the molar mass of the compound: $M = -13661.69 \text{ g/mol} - 239.0 \text{ g} = -13900.69 \text{ g/mol}$

98) determine the molar mass of the compound: $M = -13900.69 \text{ g/mol} - 239.0 \text{ g} = -14139.69 \text{ g/mol}$

99) determine the molar mass of the compound: $M = -14139.69 \text{ g/mol} - 239.0 \text{ g} = -14378.69 \text{ g/mol}$

100) determine the molar mass of the compound: $M = -14378.69 \text{ g/mol} - 239.0 \text{ g} = -14617.69 \text{ g/mol}$

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